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4 Problems and Solutions in Atomic, Nuclear and Particle Physics Solution: The Bohr radius of the hydrogen atom and the Compton wavelength of an electron are given by  $a_0 = \frac{2}{\alpha} \frac{h}{m_e c}$  and  $\lambda_c = \frac{h}{m_e c}$  respectively. Hence  $a_0 \lambda_c = \frac{2}{\alpha} \frac{h^2}{m_e c^2} = \frac{2}{\alpha} \frac{1}{137} \frac{h^2}{m_e c^2} = \frac{2}{\alpha} \frac{1}{137} \frac{1}{22}$ , where  $\frac{2}{\alpha}$  is the fine-structure constant. Hence the answer is (a). 1004 Estimate the electric field needed to pull an electron out of an atom in

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Answer:  $\rho = 737 \text{ kg/m}^3$ ;  $1 \text{ gal} = 3.788 \text{ L} = 3.788 \times 10^{-3} \text{ m}^3$ . The mass of a  $^{235}\text{U}$  nucleus is  $235m_p$ , then. (a) for 1 gallon,  $m = \rho V = (737 \text{ kg/m}^3)(3.788 \times 10^{-3} \text{ m}^3) = 2.79 \text{ kg} = 2.79 \times 10^{-3} \text{ g}$ . So,  $(1.3 \times 10^8 \text{ J}) / (2.79 \times 10^{-3} \text{ g}) = 4.7 \times 10^5 \text{ J/g}$ . (b) 1 g contains  $(1.00 \times 10^{-3} \text{ kg}) / (235m_p) = 2.55 \times 10^{21}$  nuclei.

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