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Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

Chemistry Notes for class 12 Chapter 2 Solutions

Solutions Class 12 Notes Chemistry Chapter 2 1. A solution is a homogeneous mixture of two or 9. more chemically non-reacting substances. The components of a solution generally cannot be separated by filtration, settling or centrifuging. 2. A solution may be classified as solid, liquid or a gaseous solution. 3. Solubility is defined as the [...]

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IDEAL SOLUTIONS; The solutions which obeys the Raoult's Law over the entire range of concentration are known as Ideal Solution. PROPERTIES : (i) $\Delta_{mix} H = 0$ i.e. The enthalpy of mixing of the pure component to form solution is 0. It is neither exothermic nor endothermic . (ii) $\Delta_{mix} V = 0$ i.e. There is no change in volume when 2 components are mixed to form solution.

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